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# 5-(2-Pyridyl)-1,3-dithiane-2-thione 

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The title compound, $\mathrm{C}_{9} \mathrm{H}_{9} \mathrm{NS}_{3}$, crystallizes with two molecules in the asymmetric unit. In both molecules, the dithiane-2thione rings adopt a symmetric half-boat conformation with the C atom opposite the $\mathrm{C}-\mathrm{S}_{\text {thione }}$ bond out of the plane. The pyridine ring is in an equatorial position and is twisted out of the plane of the half-boat by 82.7 (2) and $84.5(2)^{\circ}$ in the two molecules, so that the N atom is trans to the axial $\mathrm{C}-\mathrm{H}$ bond in both cases.

## Comment

We were unsuccessful in reproducing the synthesis of $S, S^{\prime}$ -[2-(2-pyridyl)trimethylene]bis( $O$-ethyl dithiocarbonate), (I) (Uneme et al., 1992). Instead, the previously unknown ringclosed product 5-(2-pyridyl)-1,3-dithiane-2-thione, (II), was recovered. The only important difference in the procedures was that we heated the heterogeneous reaction mixture for a short time.


The crystal structure of (II) has two almost identical molecules in the asymmetric unit (Fig. 1). Both the dithiane-2thione rings adopt a symmetric half-boat conformation, with the C atom opposite the $\mathrm{C}-\mathrm{S}_{\text {thione }}$ bond out of the plane defined by the other atoms; the largest deviations from the plane are 0.05 (4) $\AA$ for S 3 and 0.04 (2) $\AA$ for $\mathrm{S} 2 A$. This symmetric half-boat conformation is one of the intermediates postulated for the preferential mode of inversion for the chair form of cyclohexanone (Bucourt \& Hainaut, 1967). This conformation is also found for 1,3-dithiane-2-thione in a series
of copper complexes (Bellito et al., 1994). The bond distances (Table 1) found here $\left[\mathrm{S}=\mathrm{C} 1.658\right.$ (1) and 1.662 (1) $\AA, \mathrm{Csp}^{2}-\mathrm{S}$ 1.714 (1)-1.719 (1) $\AA$ and $\mathrm{S}-\mathrm{Csp}^{3} 1.810(1)-1.815$ (1) $\AA$ A are very similar to those in the copper complexes, although there the ligand is mono-, di- or tridentate. For both molecules, the pyridine ring is in an equatorial position and is twisted by 82.7 (2) and $84.5(2)^{\circ}$ out of the plane of the half-boat. In both molecules, the N atom is trans to the axial $\mathrm{C}-\mathrm{H}$ bond so that the molecules have approximate mirror symmetry. The two


Figure 1
View of the two molecules of the title compound showing the labelling of the non-H atoms. Displacement ellipsoids are shown at the $50 \%$ probability level and H atoms are drawn as small circles of arbitrary radii. The short $\mathrm{S} \cdots \mathrm{S}$ interaction is shown by a dashed line.
molecules are not exactly identical; superimposing the non-H atoms gives a mean-square deviation of $0.42 \AA^{2}$. The difference is mainly in the angle between the $\mathrm{C} 2 / \mathrm{C} 3 / \mathrm{C} 4$ and $\mathrm{S} 1 / \mathrm{S} 2 /$ $\mathrm{S} 3 / \mathrm{C} 1 / \mathrm{C} 2 / \mathrm{C} 4$ planes, which is $124.8(2)^{\circ}$ for molecule 1 and 120.4 (2) ${ }^{\circ}$ for molecule 2 (the molecule with suffix $A$ on the labels). This, together with a difference in the angle between $\mathrm{C} 5-\mathrm{C} 3$ and the $\mathrm{C} 2 / \mathrm{C} 3 / \mathrm{C} 4$ plane [55.5 (1) ${ }^{\circ}$ for molecule 1 and $54.0(1)^{\circ}$ for molecule 2] means that the angles between $\mathrm{C} 5-$ C 3 and the $\mathrm{S} 1 / \mathrm{S} 2 / \mathrm{S} 3 / \mathrm{C} 1 / \mathrm{C} 2 / \mathrm{C} 4$ plane differ by $6.0(2)^{\circ}$, so that in molecule 2 , the pyridine ring is bent out of plane.

The shortest intermolecular S...S contact is 3.570 (1) $\AA$ between S2 and $\mathrm{S} 3 A\left(x,-\frac{1}{2}-y, \frac{1}{2}+z\right)$. The $x$ and $y$ coordinates of the two independent molecules are related approximately by $\frac{3}{2}-x$ and $-y$, but there is no simple relation between the $z$ coordinates.

## Experimental

Triethylamine ( $3.95 \mathrm{~g}, 39 \mathrm{mmol}$ ) was added dropwise to a solution of 2-(2-pyridyl)-1,3-propanediol ( $2.40 \mathrm{~g}, 16 \mathrm{mmol}$ ) (Guanti et al., 1997) and $p$-toluenesulfonyl chloride $(6.55 \mathrm{~g}, 34 \mathrm{mmol})$ in acetonitrile $(32 \mathrm{ml})$ at 273 K over a period of ca 30 min . The mixture was kept at 279 K for 24 h and then poured into water ( 120 ml ). An oil precipitated which crystallized after removal of the water and the addition of a little ethanol. The crude product was recrystallized from ethanol, giving 3.10 g ( $43 \%$ ) of 2 -(pyridyl)trimethylene $\operatorname{bis}(p$-toluenesulfonate $) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right): \delta 2.40(6 \mathrm{H}, s), 3.46(1 \mathrm{H}$, quintet $)$,
$4.32(4 \mathrm{H}, d), 7.0-7.4(6 \mathrm{H}, m), 7.5-7.8(5 \mathrm{H}, m), 8.3-8.4(1 \mathrm{H}, m)$. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 22\left(\mathrm{CH}_{3}\right), 46(\mathrm{CH}), 68\left(\mathrm{CH}_{2}\right), 122.5($ py-C3 $), 124$ (py-C5), 128 (Ar-C3,C5), 130 (Ar-C2,C6), 132 (Ar-C4), 137 (py-C4), 134 (Ar-C1), 149.5 (py-C6), 155.5 (py-C2). A suspension of 2-(pyridyl)trimethylene bis( $p$-toluenesulfonate) ( $3.1 \mathrm{~g}, 6.9 \mathrm{mmol}$ ) and potassium $O$-ethyldithiocarbonate ( $3.18 \mathrm{~g}, 20 \mathrm{mmol}$ ) in acetonitrile ( 50 ml ) was heated under reflux for 20 min and then poured into water $(250 \mathrm{ml})$. The reaction product was extracted with $\mathrm{CHCl}_{3}(3 \times$ $100 \mathrm{ml})$ and purified by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to afford a crude product (ca 300 mg ) and a yellow oil. The crude product was crystallized from ethanol; yield: 200 mg ( $14 \%$ ) of (II) as yellow crystals. The crystals used for the structure analysis were deposited from $\mathrm{CDCl}_{3}$ (m.p. $393-394 \mathrm{~K}$ ). Analysis calculated for $\mathrm{C}_{9} \mathrm{H}_{9} \mathrm{NS}_{3}$ : C 47.54, H 3.99, N 6.16 , S $42.31 \%$; found: C 47.59 , H 4.10 , N 6.17 , S $32.30 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 3.2-3.8(5 \mathrm{H}, m), 7.2-7.4(3 \mathrm{H}, m)$, 7.6-7.8 $(1 \mathrm{H}, m), 8.5-8.6(1 \mathrm{H}, m) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 39.1\left(\mathrm{CH}_{2}\right)$, 39.2 (CH), 122.2 (py-C3), 122.7 (py-C5), 137 (py-C4), 150 (py-C6), 160 (py-C2), $220(\mathrm{C}-\mathrm{S})$; IR: $1010 \mathrm{~cm}^{-1}, \nu(\mathrm{C}=\mathrm{S})$; EIMS: $m / z 227$ $\left(M^{+}, 58\right), 194$ (23), 180 (17), 162 (11), 150 (10), 136 (26), 118 (32), 106 (100), 79 (25).

## Crystal data

## $\mathrm{C}_{9} \mathrm{H}_{9} \mathrm{NS}_{3}$

$M_{r}=227.38$
Monoclinic, $P 2_{1} / c$
$a=17.444$ (2) А
$b=9.6329$ (9) $\AA$
$c=12.680(1) \AA$
$\beta=110.245$ (2) ${ }^{\circ}$
$V=1999.1(3) \AA^{3}$
$Z=8$

## Data collection

Siemens SMART CCD diffractometer
$\omega$ rotation scans with narrow frames
Absorption correction: by integration (XPREP; Siemens, 1995)
$T_{\text {min }}=0.687, T_{\text {max }}=0.894$
25069 measured reflections
6191 independent reflections

## Refinement

Refinement on $F$
$R=0.024$
$w R=0.034$
$S=1.27$
4802 reflections
235 parameters
$D_{x}=1.511 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $\mathrm{K} \alpha$ radiation
Cell parameters from 7124
$\quad$ reflections
$\theta=2.5-30.5^{\circ}$
$\mu=0.69 \mathrm{~mm}^{-1}$
$T=120 \mathrm{~K}$
Plate, yellow
$0.58 \times 0.40 \times 0.16 \mathrm{~mm}$

4802 reflections with $I>3 \sigma(I)$
$R_{\text {int }}=0.047$
$\theta_{\text {max }}=30.5^{\circ}$
$h=-24 \rightarrow 24$
$k=-13 \rightarrow 13$
$l=-17 \rightarrow 17$
Intensity decay: none

H -atom parameters constrained
$w=1 /\left\{\left[\sigma_{\mathrm{cs}}\left(F^{2}\right)+1.03 F^{2}\right]^{1 / 2}-|F|\right\}^{2}$
$(\Delta / \sigma)_{\max }=0.001$
$\Delta \rho_{\max }=0.48(5) \mathrm{e} \AA^{-3}$
$\Delta \rho_{\text {min }}=-0.39(5)$ e $\AA^{-3}$

Table 1
Selected bond lengths ( $\AA$ ).

| S1-C1 | $1.658(1)$ | $\mathrm{S} 1 A-\mathrm{C} 1 A$ | $1.662(1)$ |
| :--- | :--- | :--- | :--- |
| S2-C1 | $1.719(1)$ | $\mathrm{S} 2 A-\mathrm{C} 1 A$ | $1.717(1)$ |
| S2-C2 | $1.815(1)$ | $\mathrm{S} 2 A-\mathrm{C} 2 A$ | $1.810(1)$ |
| S3-C1 | $1.719(1)$ | $\mathrm{S} 3 A-\mathrm{C} 1 A$ | $1.714(1)$ |
| S3-C4 | $1.812(1)$ | $\mathrm{S} 3 A-\mathrm{C} 4 A$ | $1.815(1)$ |

The H atoms were included using a riding model and were constrained to have $\mathrm{C}-\mathrm{H}=0.95 \AA$ and $U_{\text {iso }}=1.2 U_{\text {eq }}$ of their parent atom.

Data collection: SMART (Siemens, 1995); cell refinement: SAINT (Siemens, 1995); data reduction: SAINT; program(s) used to solve structure: SIR97 (Altomare et al., 1997) and KRYSTAL (Hazell, 1995); program(s) used to refine structure: modified ORFLS (Busing et al., 1962) and KRYSTAL; molecular graphics: ORTEPIII (Burnett \& Johnson, 1996) and KRYSTAL; software used to prepare material for publication: KRYSTAL.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: JZ1452). Services for accessing these data are described at the back of the journal.

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